

## Behavior of xSbNbO<sub>4</sub> additive (1-x)Ba(Zr<sub>0.2</sub>Ti<sub>0.8</sub>)O<sub>3</sub> ceramics with ferroelectric properties

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### Abstract

(1-x)Ba(Zr<sub>0.2</sub>Ti<sub>0.8</sub>)O<sub>3</sub>-xSbNbO<sub>4</sub> ceramics were synthesized by the solid state reaction method to examine the crystalline structure, and dielectric and ferroelectric properties of the samples. XRD patterns exhibited the microstructure and tetragonal phase for all samples with sintering temperature at 1150 °C. The results showed that 1 mol% SbNbO<sub>4</sub> coexisted between the tetragonal and cubic phases. The dielectric constant was increased by adding SbNbO<sub>4</sub> for 3 mol%, while the dielectric loss decreased slightly. The *P-E* loops showed normal ferroelectric and relaxor ferroelectric behavior. In addition, the good ferroelectric properties exhibited  $E_c = 2.80 \text{ kV}\cdot\text{cm}^{-1}$  and  $P_r = 7.24 \text{ }\mu\text{C}\cdot\text{cm}^{-2}$  for 4 mol% SbNbO<sub>4</sub> which originated to the Perovskite structure tetragonal phase.

**Keywords:** Microstructure; Dielectric properties; Polarizability; Solid state reaction

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### 1. Introduction

Ba(Zr<sub>0.2</sub>Ti<sub>0.8</sub>)O<sub>3</sub> is a ferroelectric material with a perovskite-type structure (ABO<sub>3</sub>) [1]. Ferroelectric materials are used for development in the electronic industry, such as sensors, dielectric amplifiers, detectors, capacitors and multilayer ceramic capacitors (MLCC) [2]. Today, ferroelectric materials are popular topics of research. This is partly due to concerns of PbO toxicity. Research on lead-free piezoelectric ceramics with excellent properties have been replaced in recent years [3]. Ba(Zr<sub>0.2</sub>Ti<sub>0.8</sub>)O<sub>3</sub> exhibit a high dielectric constant ( $\epsilon_r$ ) at about 1,400 [2 – 9], lower dielectric loss [4 – 9], a remnant polarization ( $P_r$ ) at about  $13.5 \text{ kV}\cdot\text{cm}^{-1}$  [4,6], coercive field ( $E_c$ ) of about 2.9  $\text{kV cm}^{-1}$  [6], density of about  $6.02 \text{ g}\cdot\text{cm}^{-3}$  [4, 10] and grain size at about  $\sim 1 \text{ }\mu\text{m}$  [4 – 10]. However, the properties of SbNbO<sub>4</sub> are inferior to the Ba(Zr<sub>0.2</sub>Ti<sub>0.8</sub>)O<sub>3</sub> (BZT) system. Several researchers have studied additions to composites, such as xBaZrO<sub>3</sub>-(1-x)BaTiO<sub>3</sub> [3], Ba(Zr<sub>x</sub>Ti<sub>1-x</sub>)O<sub>3</sub> [4], xCa-xBaTiO<sub>3</sub> [5], xAl-xBaTiO<sub>3</sub> [6] and BZT15/85 [10]. Recently, fine grains of approximately 1  $\mu\text{m}$  have been reported as the piezoelectric properties for the Ba(Zr<sub>0.2</sub>Ti<sub>0.8</sub>)O<sub>3</sub> system. Thus, for the property enhancements of the microstructure to be optimal, the synthesized method is required. Also, importantly, the dopant could greatly support the piezoelectric properties. Among dopant, SbNbO<sub>4</sub> is a good material because it is abundant and inexpensive [9]. It also has good piezoelectric properties, with a decrease in the sintering temperature. SbNbO<sub>4</sub> is one of the materials that exhibits semiconducting and piezoelectric properties [9]. Recently, it has been used in a wide range to enhance the piezoelectric properties such as  $d_{33}$  and  $k_p$  included grain growth [9, 10].

Thus, in this paper, we present the synthesized  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-}x\text{SbNbO}_4$  composite that was conducted under different conditions by the solid state reaction method. The effects of adding  $\text{SbNbO}_4$ , including the physical and ferroelectric properties for enhanced ferroelectric properties, were systematically studied.

## 2. Materials and methods

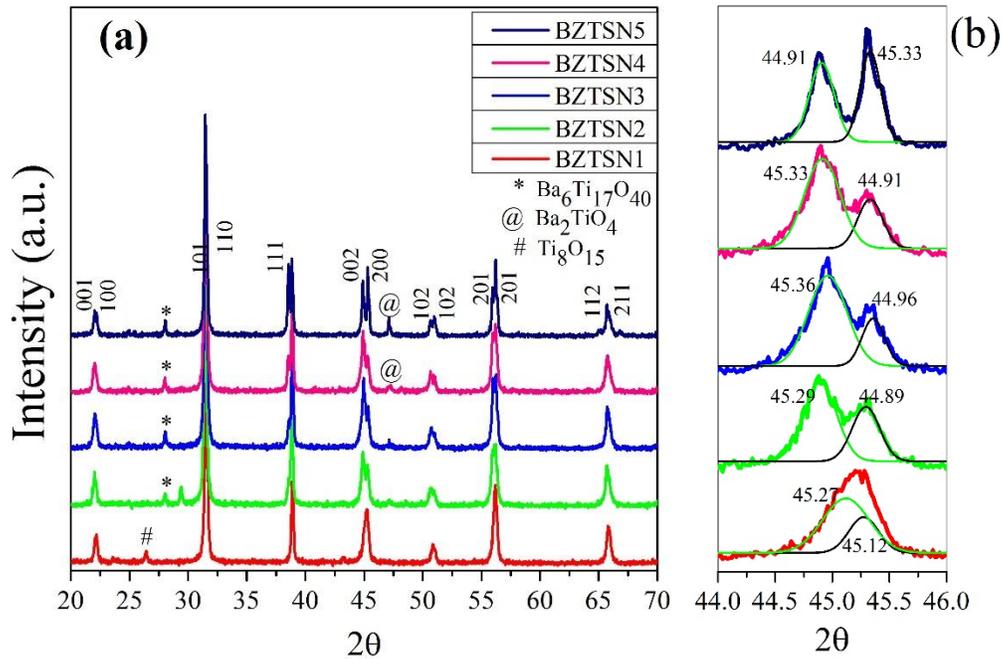
$(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-}x\text{SbNbO}_4$  systems were synthesized by the solid state reaction method with compositions of  $x = 1, 2, 3, 4$  and  $5$  mol%. The high-purity precursor powder [composed of  $\text{BaCO}_3$  (99%),  $\text{ZrO}_3$  (99%),  $\text{TiO}_2$  (99%),  $\text{Sb}_2\text{O}_5$ (98%), and  $\text{Nb}_2\text{O}_3$ (99.9%)] was used for the formula of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-}x\text{SbNbO}_4$ . Both  $\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3$  and  $\text{SbNbO}_4$  were synthesized with the separated term in which  $\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3$  and  $\text{SbNbO}_4$  were mixed at a ratio of  $\text{Ba}:(\text{Zr}:\text{Ti}) = 1:(0.5:0.5)$  and  $\text{Sb}:\text{Nb} = 1:1$  into a zirconia ball, milled in deionized water for 24 h, and then volatilized. Then, they were calcined at  $830^\circ\text{C}$  for 4 h and subsequently mashed. The powders were obtained by varying the fraction  $x$  as  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-}x\text{SbNbO}_4$ . After that, the compounds were mixed into a zirconia ball milled in deionized water for 24 h, and then the volatilization. These powers were calcined at  $830^\circ\text{C}$  for 4 h and mashed. The composition of the  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-}x\text{SbNbO}_4$  was mixed PVA and uniaxially pressurized under 320 MPa into the mold to disc samples. The condition of sintering temperature at  $1,150^\circ\text{C}$  with a heating rate of  $10^\circ\text{C}\cdot\text{min}^{-1}$  under atmospheric for 2 h, then soaked and cooled to room temperature. The investigation used crystal structures with x-ray diffraction technique (Shimadzu XRD-6100), where the diffraction intensity was measured in a range between  $20^\circ$  to  $70^\circ$  with a step-up of  $0.02^\circ$ . ferroelectric properties, disc samples were obtained by coating the two main surfaces with silver paint for poling treatment, hysteresis loop and dielectric constant measurement. The dielectric measurement was conducted at 30 to  $300^\circ\text{C}$ , using a Chen Hwa 1061 LCR-Meter. The capacitance was determined by  $\epsilon_r = \frac{Ct}{\epsilon_0 A}$  ( $\epsilon_r$  is dielectric constant,  $C$  is capacitance,  $t$  is thickness,  $\epsilon_0$  is permittivity of a vacuum,  $A$  is area).

## 3. Results and Discussion

### *Physical Properties*

The XRD patterns of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-}x\text{SbNbO}_4$  with the composition of 1, 2, 3, 4 and 5 mol% are shown in Fig. 1. The tetragonal (P4mm) phase structure of all samples is seen in Fig. 1(a). It is evident that the major phase is  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-}x\text{SbNbO}_4$ , following the existing of  $\text{Ba}_6\text{Ti}_{17}\text{O}_{40}$  and  $\text{BaTi}_2\text{O}_5$  phases (showing as a cubic phase). The diffraction peaks of the tetragonal system (P4mm) could be indexed as perovskite structures in the XRD graphs [1, 3]. The samples showed the ferroelectric perovskite phases coexisted between the tetragonal to pseudo (cubic) phases. Fig. 1(b), The splitting peaks between  $(002)_T$  and  $(200)_T$  indicated the tetragonality of  $c/a$  with zoomed  $2\theta$  rang  $44 - 46^\circ$ , conformed by the percentage of tetragonality, which was observed by the change of  $(002)_T$  and  $(200)_T$  peaks in Fig. 1 (b). Thus, the  $\text{SbNbO}_4$  content increased with influence to lattice parameter  $a$  increases,  $c$  decreases; hence, the tetragonal ( $c/a$ ) distortion increased, as shown in Fig. 2. The abnormal tetragonality was lowest in the 1 mol% sample because the  $(002)_T$  and  $(200)_T$  peaks appear with coexisting as shown in Fig. 1(b) [1, 3, 9]. In contrast, the 4 mol% sample obtained high tetragonality were split at  $(200)_T$ . Thus, the results of XRD assume that the added 2 mol% may be near Morphotropic Phase Boundaries (MPB phase), which indicates that an overlap of the structures has a negative impact on symmetry, with good ferroelectric properties [12]. When  $\text{SbNbO}_4$  content was further increased, there were visible affirm with increases of  $(002)_T$  and  $(200)_T$ , supplementary in percentage tetragonality. At more

than 3 mol% content, the tetragonal phases became dominant in the  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$  as  $x$  increased. Generally, the incorporation of elements for the site in the  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$  perovskite structure can be predicted by considering the corresponding ionic radii and the Goldschmidt tolerance factor  $t = \frac{r_A+r_O}{\sqrt{2}(r_B+r_O)}$  [15]. The tolerance factor obtained incorporation the ionic radii of  $\text{Ba}^{2+}$  (1.35 Å),  $\text{Ti}^{4+}$  (0.745 Å),  $\text{Zr}^{2+}$  (0.74 Å) and  $\text{O}^{2-}$  (1.4 Å), the  $t$  value is 0.906 of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$ . Therefore, the substitution of  $\text{Zr}^{2+}$  may occur more in B site [9] that the  $t$  of B-site (Ti) substitution in this doping is close to 1; therefore, B site substituting is more likely to occur).



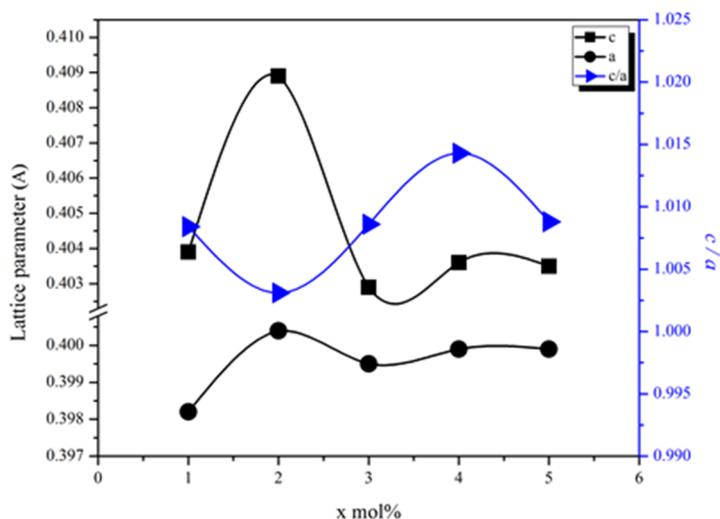
**Fig. 1** (a) XRD patterns of the  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$  (BZTSN) with  $x=1, 2, 3, 4$  and  $5$  mol% and sintering at  $1150\text{ }^\circ\text{C}$ . (b) Enlarged XRD graph of zoom range  $44 - 46^\circ$ .

**Table 1** ferroelectric properties of the  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$  (BZTSN)ceramics.

| samples | Perovskite phase (%) | Dielectric constant ( $\epsilon_r$ ) | Temperature max ( $^\circ\text{C}$ ) | $P_r$ ( $\mu\text{C}\cdot\text{cm}^{-2}$ ) | ( $E_c$ ) ( $\text{kV}\cdot\text{cm}^{-1}$ ) |
|---------|----------------------|--------------------------------------|--------------------------------------|--------------------------------------------|----------------------------------------------|
| BZTSN1  | 95.30                | 722                                  | 176                                  | 13.38                                      | 7.24                                         |
| BZTSN2  | 72.68                | 829                                  | 180                                  | 1.23                                       | 1.00                                         |
| BZTSN3  | 86.28                | 767                                  | 184                                  | 7.63                                       | 5.05                                         |
| BZTSN4  | 84.72                | 1038                                 | 188                                  | 7.24                                       | 2.80                                         |
| BZTSN5  | 91.83                | 908                                  | 188                                  | 12.89                                      | 4.68                                         |

Table 1 shows the ferroelectric properties of the  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$  ceramics. It was found that the perovskite phases reached a maximum of 1 mol% (about 95.30), with opine the carbon atoms have lower concentration which non-perovskite structure [1,9,10]. Moreover, as the lattice parameter decreases,  $c$  increases, which also increases tetragonality ( $c/a$ ). The tetragonality  $c/a$  of these ceramics obtained decreased when the  $\text{SbNbO}_4$  content increased, but the  $\text{Ba}^{2+}$  and  $\text{Ti}^{4+}$  cations were also enhanced, as shown in Fig. 2. This  $c/a$  ratio changed when the composite concentration increased. In the 2 mol% sample, the  $c/a$  ratio increased at 1.0159 and then the  $c/a$  ratio decreased at 1.0079 added composites ample to 3 mol.%. Therefore, the Bragg angle of appeared splitting several peaks to percentage perovskite phase increases was more than 85%, as

shown in Table 1. The relations of ferroelectric properties were  $\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$  ceramics, with sintering at 1,150 °C for 2 h. The dielectric properties exhibit with a direct relationship between  $\epsilon_r$  on  $T_{max}$  which increased with the addition of  $\text{SbNbO}_4$  content; meanwhile the  $E_c$  present increased as listed in Table 1. It was observed that the 4 mol% sample shows significantly different  $\epsilon_r$  and  $E_c$ , regarding with  $\text{Sb}^{2+}$  and  $\text{Nb}^{2+}$  ions affecting the dielectric properties, within the conductivity of behavioral ceramics.  $\text{SbNbO}_4$  content into samples ceramics was attempt modified on dielectric property,  $\text{Sb}^{2+}$  and  $\text{Nb}^{2+}$  seriously helpful to adjust electrical properties [9]. Interestingly, with 5 mol%  $\text{SbNbO}_4$ , the dielectric loss increased remarkably, and the dielectric constant decreased. However, a report corresponding to Soma Chattopadhyay *et al.*, stated that Sb and Nb have good dielectric properties [9]; the dielectric loss increased rapidly, but the dielectric constant decreased. From  $P$ - $E$  hysteresis loops for the  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$  ceramics with different  $\text{SbNbO}_4$  contents, the evolution data of  $P_r$  and  $E_c$  as functions of  $\text{SbNbO}_4$  content are listed in Table 1.  $P_r$  and  $E_c$  reach their maximum with the increase of 1 mol% and then decreased rapidly with an increase of 2 mol%. In further addition,  $P_r$  obtains further increases value about 7.63, 7.24 and 12.89  $\text{kV}\cdot\text{cm}^{-1}$ . Fig. 2 for 3, 4 and 5 mol%, respectively. For the  $\text{SbNbO}_4$  addition, these samples were sintered with an increase in grain growth, since it has been reported that grain size reduces the orientation of polarization along electric fields [9], thus ferroelectric properties have been influenced untroubled. Although 2 mol% exhibited significantly high values, with character  $P$ - $E$  loops showing normal ferroelectric behavior, regularly exhibited relaxor ferroelectric behavior for BZT [10] indicated a complex perovskite phase. Table 1 shows the remnant polarization ( $P_r$ ) and coercive field ( $E_c$ ) for the  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$  at different mol% content. 5 mol% showed a higher value at around 12.11  $\text{kV}\cdot\text{cm}^{-1}$ , while low the at 3.80  $\text{kV}\cdot\text{cm}^{-1}$ . Moreover, their samples revealed good values with 2.80 and 4.68  $\text{kV}\cdot\text{cm}^{-1}$  for 4 and 5 mol%, respectively. This result indicated that the ceramic obtain expedient to pole and effect has higher piezoelectric properties.

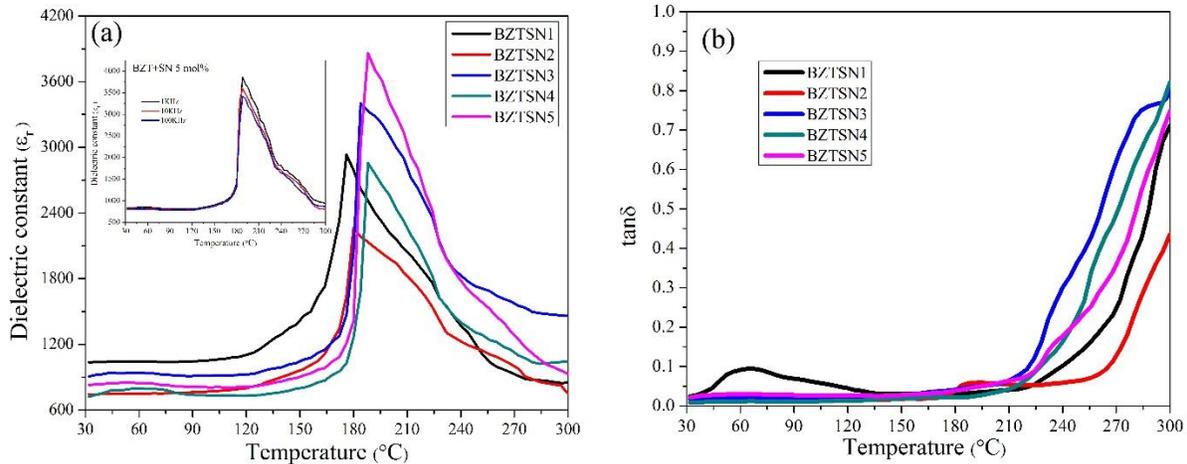


**Fig. 2** The lattice parameters of the  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$  ceramics.

### Dielectric Properties

The dielectric constant and dielectric loss of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$  adding x with 1, 2, 3, 4 and 5 mol% are shown in Fig. 3. The dielectric behavior is favored to explain the switching domain in micro and nano regions which affect phase transition at different temperatures. Dielectric constant ( $\epsilon_r$ ) and dielectric loss ( $\tan\delta$ ) as a function of temperature at various frequencies of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{-xSbNbO}_4$  have been obtained. The maximum dielectric constant at

temperature ( $T_{max}$ ) ranged from 92 – 200°C for these samples, as seen in in Table 1. Fig. 3 shows the increase of added  $\text{SbNbO}_4$  content increasing, because of higher capacitance for  $\text{SbNbO}_4$ . But, 2 mol%  $\text{SbNbO}_4$  has been the  $\epsilon_r$  leaked higher at frequency 1 kHz. Meanwhile,  $x = 3$  mol% decreased in this high temperature. Besides,  $\text{SbNbO}_4$  the broaden peak of  $\epsilon_r$  exhibited dispersion, which raised with increasing temperatures at the various frequencies, as given in Fig. 3. The was found raised because pinning domain occurred by  $\text{Zr}^{2+}$  such domain wall that could not be switching. While added both the  $\text{SbNbO}_4$ , the broaden peak of exhibited distribution which raised when the increasing temperature at a various frequency as give in Fig. 3(a). The of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3-x\text{SbNbO}_4$  4 mol% was found shift to have the maximum dielectric constant at temperature ( $T_{max}$ ) from 30 – 300 °C with 176, 180, 184, 188 and 188 °C for  $x = 1, 2, 3, 4$  and 5 mol%, respectively [see in Fig. 3]. Generally, the BZT base showed a maximum dielectric constant of about 123 °C [1, 2, 9]. All samples exhibited a diffuse ferroelectric phase transition with a transition temperature that caused the substitution of  $\text{SbNbO}_4$  into the O site and  $\text{Zr}^{2+}$  ions into the B site leading to disorder of complex perovskite structure.



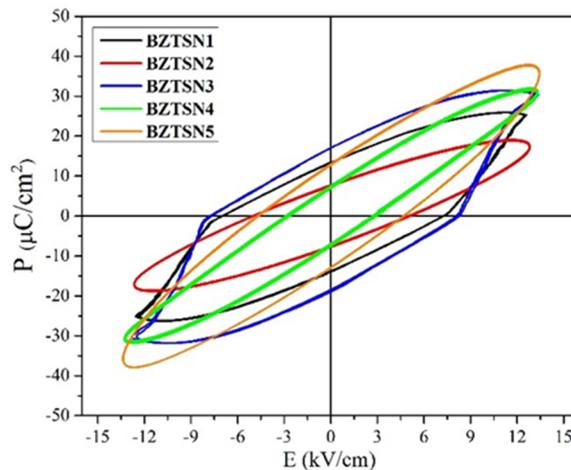
**Fig. 3** Dielectric Properties of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3-x\text{SbNbO}_4$  sintering at 1,150°C.

### Ferroelectrics

The measurement of the polarization-electric field ( $P$ - $E$ ) hysteresis loops was performed to examine the ferroelectric properties of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3-x\text{SbNbO}_4$  ceramics. Fig. 4 displays the different  $P$ - $E$  hysteresis loop shapes which can be used to indicate ferroelectric behavior with ceramic composition at room temperature.  $P$ - $E$  loop of 2 mol%  $\text{SbNbO}_4$  ceramic was squareness given in Fig. 4. The coercive field ( $E_c$ ), remnant polarization ( $P_r$ ), spontaneous polarization ( $P_s$ ) sharpness of points, and squareness of loop indicated normal ferroelectric behavior. Moreover, it exhibits increasing the domain size and dipoles. The ferroelectric (FE) state displays disruption by the zirconium-titanate (Zr, Ti), which may be a consequence of FE-AFE transformation at high temperatures [13, 15]. While 2 mol%  $\text{SbNbO}_4$  gave as a flat loop with rapidly lower for  $P_r$ , and  $E_c$ . These results decreased the ferroelectric domain size, which indicates nanodomains in relaxor ferroelectrics decreases the spontaneous polarization. Nonetheless, with further addition, relaxor ferroelectric behavior has emerged for 3, 4 and 5 mol%. These include ferroelectric domain size increasing to microdomains along with  $E_c$ ,  $P_r$  and  $P_s$ , as depicted in Fig. 4.

Fig. 4 shows characteristics of the polarization-electric field ( $P$ - $E$ ) hysteresis loops of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3-x\text{SbNbO}_4$  ceramics at room temperature. All samples possess ferroelectric behavior from the  $P$ - $E$  hysteresis loop. The characteristic of  $P$ - $E$  hysteresis loops exhibited relaxor ferroelectric behavior, indicating to the domain size. In the 3 mol% sample, it observing attempt

in change ferroelectric behavior, cause of  $\text{SbNbO}_4$  substituted due to the B-site of  $\text{Ti}^{4+}$ . Therefore, the behavior decreased the spontaneous polarization ( $P_s$ ), as displayed in Fig. 4. Thus, the domain size rapidly decreased and then increased with the content of 4 mol%. Fig. 4, illustrates several the remnant polarizations ( $P_r$ ) and coercive fields for the  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3-x\text{SbNbO}_4$  at different contents. The BZT shows a higher value at around  $12.11 \text{ kV}\cdot\text{cm}^{-1}$ , while low at  $3.8 \text{ kV}\cdot\text{cm}^{-1}$ . With 4 and 5 mol% added, the samples exhibited high values of  $14.86$  and  $15.49 \text{ kV}\cdot\text{cm}^{-1}$  Fig. 4, respectively. Moreover, their samples obtained good values with  $1.79$  and  $2.13 \text{ kV}\cdot\text{cm}^{-1}$  for 4 and 5 mol% content, respectively. This result indicates that the ceramic obtain expedient to pole and effect has higher piezoelectric properties.



**Fig. 4**  $P$ - $E$  hysteresis loop of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3-x\text{SbNbO}_4$  sintering at  $1,150 \text{ }^\circ\text{C}$ .

#### 4. Conclusion

The temperature of the dielectric constant maximum ( $T_{max}$ ) shifted from  $180 \text{ }^\circ\text{C}$  to the  $188 \text{ }^\circ\text{C}$  with increasing  $\text{SbNbO}_4$  content. While the dielectric constant ( $\epsilon_r$ ) was further increased at the  $\text{SbNbO}_4$  contents of 2, 3, and 4 mol%, it decreased with 5 mol%  $\text{SbNbO}_4$ . During the addition of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3-x\text{SbNbO}_4$ , when a further amount of  $\text{SbNbO}_4$  was added to the BZT, the dielectric loss decreased because of capacitance in the  $\text{SbNbO}_4$  content, but rose at higher temperatures. In  $\text{SbNbO}_4$  addition, the dielectric property was fabricated electrical conductance of  $\text{SbNbO}_4$ . The dielectric constants slightly decreased, while the dielectric loss increased stability with increased temperature. Moreover, with the addition of  $(1-x)\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3-x\text{SbNbO}_4$ , the  $P$ - $E$  hysteresis loop became narrower as the  $\text{SbNbO}_4$  content increased, but displayed normal and relaxor ferroelectric behavior. The formation of  $\text{SbNbO}_4$  has been enlarged with the reduction of the oxygen vacancy. The ferroelectric properties show a higher remnant polarization ( $P_r$ ) and coercive field ( $E_c$ ) at about  $13.38 \text{ } \mu\text{C}\cdot\text{cm}^{-2}$  and  $7.24 \text{ kV}\cdot\text{cm}^{-1}$  for 1 mol%. With the increasing  $\text{SbNbO}_4$  content,  $P_r$  and  $E_c$  decreased. The ferroelectric properties of samples exhibited relaxor ferroelectric behavior at room temperature. The electrical properties were good  $P_r$  and  $E_c$  for 4 and 5 mol% were considered good.

#### 5. Suggestions

The synthesis should be prepared in an inert gas atmosphere. It had the reducing reaction on oxygen for firing  $\text{SbNbO}_4$ .

#### 6. Acknowledgement

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