



Stability of alginate encapsulation beads for microalgae cultivation

Narunat Sewiwat¹⁾, Tim C. Keener²⁾ and Thunyalux Ratpukdi^{*1)}

¹⁾Department of Environmental Engineering, Faculty of Engineering, Khon Kaen University, Khon Kaen 40002, Thailand.

²⁾Department of Biomedical, Chemical and Environmental Engineering, College of Engineering and Applied Science, University Of Cincinnati, Cincinnati, OH, 45221 U.S.A.

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Abstract

This research investigated the preparation of calcium alginate beads by encapsulation technique for algae cultivation for biofuel. The effects of calcium chloride (CaCl_2) to sodium alginate ratios of 3:0.5, 4:0.5, and 5:0.5 were tested for physical, chemical and biological stabilities. Solutions at pH of 2, 4, 7, 9, and, 11 were tested. The effects of different medium including tap water, fertilizer water (FER), effluent anaerobic digester 5% dilution (EAD5) and 10% dilution (EAD10) with tap water, a raw domestic sewage (RSW) were evaluated. The results show that the bead can with stand most tested conditions except for biological stability test that use effluent from anaerobic digester (60-80% stable). Further study on algae growth in those conditions (CaCl_2 : SA ratio) is required.

Keywords: Algae, Wastewater from anaerobic digester, Growth rate, Calcium alginate, Cell encapsulation technique

1. Introduction

Fuel production from algae oil has become interested to public during the past decades as it is one of alternatives of renewable and sustainable energy. Oil from algae has advantages over terrestrial crops because the yield per unit area (95,000 L/ha) is higher than other terrestrial crops such as palm oil (5950 L/ha) [1]. Also it does not affect the crops that are made for food supply.

Algae has requires solar energy in photosynthesis and help decrease carbon dioxide as a greenhouse gas to create new algae cell. Another factor that affects the algae grow this the nutrient that use synthetic chemicals will increase the costs. Algae harvesting is one of the important aspect that also affects the overall algae biofuel production. Since the size of algae is relatively small and not easily settled, separating them could add the cost as well [2].

Cell immobilization and cell encapsulation are cell entrapment techniques that can be used to help with algae separation. Polymer substances (such as calcium alginate) and hardening agents (such as CaCl_2) are used to form the hydrogel bead where algae can grow inside [2]. As a result, the algae beads can be easily separated from water and later extracted for lipid. The conceptual design of algae beads cultivation in the reactor is shown in Figure 1. To prepare the alginate beads for algae cultivation for biofuel, the ratio of chemicals used to create the beads must be determined for stability in the conditions supporting the growth of algae. The strength of the bead itself should be tested without algae cell prior to cell addition. Therefore, this research aimed to determine the ratio of mixing chemicals to form encapsulated

alginate beads. Also, testing the physical, chemical and biological stability of alginate beads formed under different ratio of chemicals.

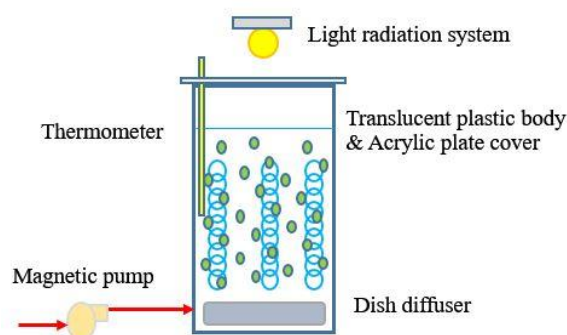


Figure 1 Conceptual design of encapsulated algae reactor

2. Materials and methods

2.1 Chemicals for alginate bead encapsulation

Alginate capsules (beads) are prepared from sodium alginate (SA) (CAS No. 9005-38-3, Sigma-Aldrich, MO, USA) solution (0.5 %w/v) and calcium chloride (Sigma-Aldrich, MO, USA) solution. To form alginate beads, CaCl_2 solution was dropped into stirred sodium alginate solution (Figure 2). Once the beads were formed, they were strained and soak in 2% w/v CaCl_2 for 2 hours for hardening.

*Corresponding author. Tel.: +6680 469 2440
Email address: thunyalux@kku.ac.th
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Three concentrations of CaCl_2 solution (3, 4, and 5% w/v) were used to prepare alginate beads.

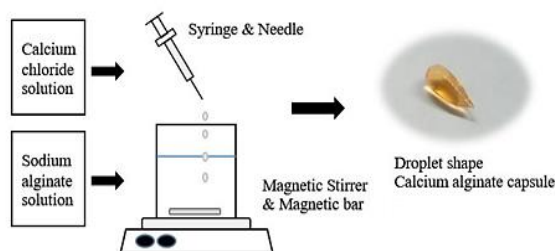


Figure 2 Alginate bead encapsulation process

2.2 Experimental setup and design

The experiments for testing alginate bead stability consist of 3 tests including physical, chemical, and biological stabilities. The tests were performed to determine which bead preparation condition could withstand the environmental condition that might be exposed during the algae cultivation. It was noted that this experiment is only a preliminary test for the alginate beads preparation. Therefore, no algae cell was introduced in all of these tests. The stability of the beads with algae cell will be further investigated (not in this study) after the best condition is obtained.

2.2.1 Physical stability test

This test was performed to determine the effect of shear and lateral force between calcium alginate beads in the reactor. Twenty alginate beads and 15 of glass beads were placed in 50 mL of deionized (DI) water in a 250-mL Erlenmeyer flask. The flask was then shaken at 100 rpm for 7 days. The alginate beads were observed daily to determine the breakage or damage. The experiments were triplicated.

2.2.2 Chemical stability test

Chemical stability test consists of two parts; 1) the effect of pH and 2) the effect of medium. For the effect of pH, solution pH of 2, 4, 7, 9, and 11 were investigated. Solution pH 2 was prepared by dissolving 7 g $\text{KHC}_4\text{H}_4\text{O}_6$ in 1 L of DI water and adjusted to pH 2 using 1 N H_2SO_4 . Solution of pH 4 was prepared from 10.12 g $\text{KHC}_8\text{H}_4\text{O}_4$ in 1 L of DI water. Solution of pH 7 was the DI water and adjusted pH by 1 N H_2SO_4 or 1 N NaOH . Solution of pH 9 was prepared from 3.80 g $\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O}$ in 1 L of DI water. Solution of pH 11 was made of 2.092 g NaHCO_3 + 2.640 g Na_2CO_3 in 1 L of DI water. Then pH was adjusted by 1 N NaOH . The alginate beads were soaked in solutions for 1 week. The alginate beads were observed daily to determine the breakage or damage. The experiments were triplicated. For the effect of medium, tap water (TW), fertilizer water (FER), effluent anaerobic digester 5% dilution (EAD5) and 10% dilution (EAD10) with tap water, a raw domestic sewage (RSW) were used since these media could be the potential substrate sources of algae cultivation. FER was prepared from commercial fertilizers: 46% urea and $\text{N-P}_2\text{O}_5\text{-K}_2\text{O}$ (8-24-24) [3]. One liter of prepared medium contains 56 mg N, 12.4 mg P and 14.8 mg K.

2.2.3 Biological stability test

The effect of biological degradation potential of bacteria in an aqueous medium was tested. A series of experiments were conducted by two separate sets of sterile and non-sterile EAD10 (will be used to culture algae). EAD10 medium was sterilized by autoclaving at 121°C for 105 min. The beads were shaken incubation for 1 week and then observed for breakage or damage. The experiments were duplicated.

3. Results

3.1 Physical stability

From the period of 1 week incubation shaking, it was found that the impact from shear force model was very little. No capsule alginate bead was damaged in all ratios of CaCl_2 :SA (Figure 3a).

3.2 pH stability

The calcium alginate beads were immersed in solution pH of 2, 7, 9 and 11 for a period of one week. During this time, no damage or breakage was observed except at pH 11 (Figure 3b).

3.3 Effect of medium

The effect of medium on stability of the calcium alginate beads is shown in Figure 4. Results showed that the effect of medium to alginate capsules was low for TW, FER and RSW. The medium that caused beads damage were EAD5 and EAD10. Bead stability in EAD5 and EAD10 ranged from 60-80%.

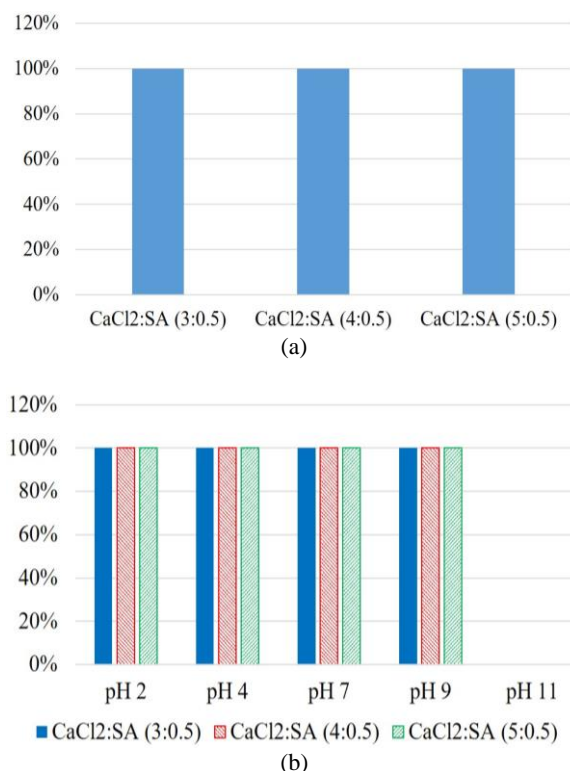


Figure 3 Physical (a) and pH (b) stability of calcium alginate beads

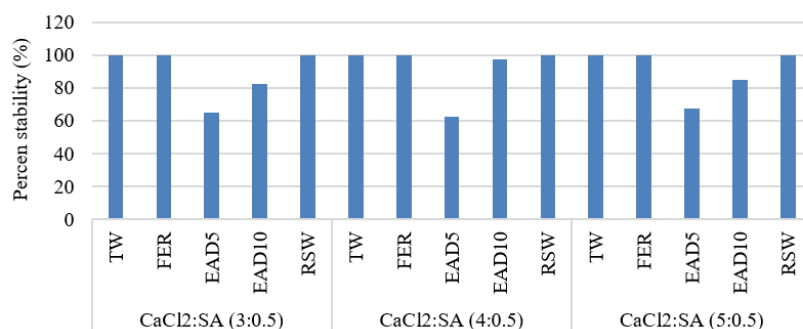


Figure 4 Effect of medium on stability of calcium alginate beads

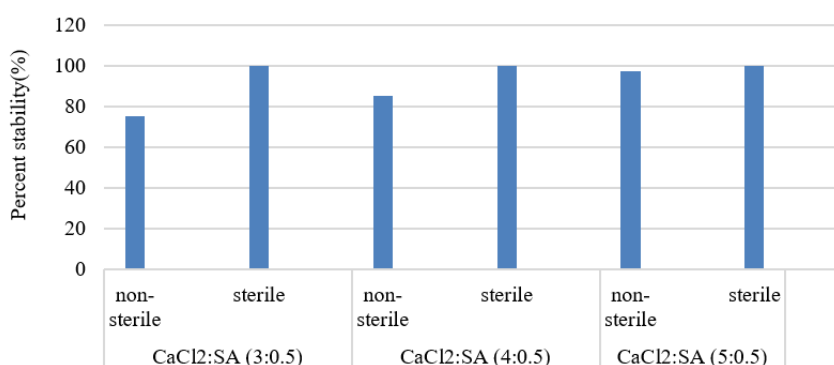


Figure 5 Biological stability of calcium alginate bead

3.4 Effect of biological degradation

From results in section 3.3, it has shown that EAD5 and EAD10 media affected calcium alginate capsules (Figure 5). It was speculated that biological might be play role in beads stability. The results show that the non-sterile EAD10 has microorganism that can damage the beads while the bead soaked in sterile EAD10 for 7 days has no damage.

4. Discussion

From the physical, chemical and biological stability tests, it was found that the calcium alginate beads were quite stable in all tested conditions except for EAD medium. The biological stability test proved that the damage of the beads could be from biological activity of microorganism in effluent of anaerobic digester. Higher CaCl₂ concentration tended to provide better stability under EAD (Figure 5). This suggested that one must be careful to use capsule algae using EAD as source of nutrients.

5. Conclusions

From this study, it was found that the calcium alginate beads prepared by an encapsulation technique have ability to withstand environmental condition for algae cultivation. Biological activity of indigenous organism could play a vital role in bead stability. The ratio of CaCl₂ to SA is important to biological stability as it increases more crosslink structure inside the bead. To determine which condition is suitable for algae cultivation and harvesting, further study of the algae growth in the beads needed to be conducted.

6. Acknowledgements

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