



Review Article

The Role of Bacteria and Factors Affecting the Biodegradation of Pollutants in the Biofilter Reactor: A Review

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Abstract

The findings showed that several types of bacteria can help biodegrade organic pollutants, such as *Comamonadacea*, which can biodegrade volatile fatty acids and aromatic compounds. *Proteobacteria*, *Bacteroidetes*, and *Actinobacteria* can biodegrade ammonium. Burkholderiales can biodegrade ferric ions and hydrogen. *Comamonas testosteroni* is able to biodegrade nitrates. *Pseudomonas taiwanensis*, *Acinetobacter guillouiae*, and *Klebsiella pneumoniae* can reduce copper, chromium and zinc levels. *Azolla* biomass reduces strontium. *Rhodospirillum* sp. can biodegrade cadmium, mercury, lead, and nickel. *Gallionella ferruginea* and *Leptothrix* sp. can biodegrade arsenic and manganese. *Gracilaria* sp. can biodegrade aluminum, chromium, and zinc. *Desulfovibrio* sp. can biodegrade copper, zinc, nickel, iron, and arsenic. *Thiomonas* sp. can biodegrade arsenic and iron. *Thauera selenatis* can biodegrade copper, zinc, cadmium, nickel, lead, cobalt, chromium, and mercury. *Thiobacillus thiooxidans* can degrade both zinc and copper. *Sargassum filipendula* biodegrades copper and nickel. Meanwhile, in the findings of the factors that affected biofiltration, it was identified that there were four that played a significant role such as temperature, dissolved oxygen, hydraulic retention time, organic loading rate, biological organisms, and supply nutrients. In conclusion, several types of bacteria grow and help biodegrade in biofilter reactors. This is inseparable from the supporting factors that increase the efficiency of pollutant reduction in biofilter reactors.

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Introduction

Water scarcity that occurs throughout the world is caused by the depletion of freshwater sources due to liquid waste pollution, which mostly occurs in poor and developing countries. However, several developed countries have reported similar problems [1–2]. Many recent cases of eutrophication have had detrimental impacts on the environment and have received tremendous attention worldwide [3]. The increasing problem of lack of clean water and polluted sources has increased the paradigm of renewable water treatment by attempting

to treat water directly from the source [4], reuse [5], and decentralization [6]. Therefore, it does not pollute the environment. Wastewater treatment techniques, both physical, chemical, and biological, have been introduced to prevent damage to clean water sources from harmful pollutant contamination. Physicochemical treatments include coagulation-flocculation, aeration, chemical oxidation, filtration, and biofiltration [7–9]. Although chemical processing has some drawbacks, such as high sludge production, the relatively high investment capital [10] does not make this processing difficult to implement.

Processing using biofiltration is indeed more preferred than other biological treatments that have been introduced, this is of course because the potential for reducing pollutants produced is quite high compared to the others, however, processing using biofiltration is still used as a third or tertiary treatment which is capable of significantly reducing pollutants in wastewater [11]. Processing with biofiltration involves the utilization of bacteria as a breakdown of pollutant elements that enter the liquid waste; later, these bacteria will be mobilized into the biofilm that has been formed, and the presence of biofilm and bacteria in the media will attract pollutants, which are in charge of breaking down pollutant elements into elements that are safer to release into the environment [12]. Several studies have shown that biofiltration can also be used to reduce pollutants in industrial gas waste and eliminate odor disturbances that may arise from gases containing harmful pollutants [13–17]. This is because biofiltration is known can remove volatile organic compounds [18].

In several municipal WWTP, biofilters were applied to remove nitrogenous pollutants and organic matter that are harmful to the environment. This is because the findings show that biofilters can be used to treat various types of pollutant problems in conventional drinking water treatment [19], gas and oil [20], raw waste [20–21], domestic wastewater [22–23], and groundwater [24]. Several studies have also developed biofilter operating conditions, including differential dissolved oxygen concentration, temperature, and filter media, which in general can affect the ability of biofilters to reduce pollutants in wastewater [25–27]. The layer of the biofilter media is very important for understanding the number of microbial structures that appear during the operation of the biofilter. Fungi, microbial activity, and bacterial viability greatly determine the effectiveness of a biofiltration reactor for reducing existing pollutants [28]. To maximize the formation of biofilms on the media, it is important to pay attention to the type of media used in the biofiltration reactor because it greatly affects the formation of aerobic and anaerobic zones, which can affect the distribution of dissolved oxygen in wastewater. [13]. However, many bio-filtration designers currently focus only on the absorption properties of biofilters rather than the formation of bacterial colonies, which will later break down pollutant elements into elements that are harmless to the environment [29–32]. The lack of information about the performance of biofilter reactors on technical and semi-technical scales is an obstacle in the development of this science because most of the previous research was carried out at the laboratory scale [15, 17, 28, 30, 33].

Energy saving is one of the reasons why many biological treatment processes are carried out compared to others, and flexibility, environmental friendliness, and lower costs also encourage other researchers to be more active in researching biological waste treatment. [10, 34–43]. Several studies related to biofilters are also interested in developing alternative media to further improve the performance efficiency of biofilms and bacteria, including wood, straw, and other fibers [44–46]. This was done to replace media that are often used, such as sand and gravel. Many pollutants in wastewater are known to occur in the scrubber system, which are then broken down by bacteria into compounds that are safer to be discharged into the environment [14, 47–48]. In the nitrification process, the microbial population in the biofilm is considered a nitrifier community whose function is to convert nitrite compounds into gases that are safely disposed of in the environment [49]. In this process, the nitrification process involves autotrophic bacteria, which convert ammonium to nitrite, and bacteria convert nitrile to nitrate [50].

Although research [51–55] has discussed several information of biofilter information have been discussed, a review of the role of microorganisms and factors that influence the effectiveness of biofiltration has not been found. The main objective was to answer the following questions: 1. Which bacteria appear during biofiltration?, 2. What factors affect microorganism-assisted biofiltration?, and 3. How does the abundance of bacteria occur in a medium biofilter? This article specifically examines the above and explains it in detail by presenting data and tables from previous studies. The novelty of this research is that there has never been a study that thoroughly discusses the types of bacteria, factors that influence bacteria to appear in biofilms, and how the abundance of bacteria can occur in biofilters. In this review, article data were developed from reputable sources from SCOPUS, EBSCO, and Proquest. The data obtained were then poured into tables and analyzed in depth to find links to the research questions.

Material and method

Supporting data for this review were obtained from articles that had been published by previous studies of reputable categories, such as Scopus, EBSCO, and Proquest. These data were then synthesized based on research needs by dividing them into several parameters that will be discussed, namely, the types of microorganisms that assist in the biofiltration process and the pollutants removed, and how the temperature factor, organic loading rate, dissolved oxygen, and retention time affect the biofiltration process in reducing

pollutants in liquid waste. Data that have been separated based on parameters are then entered into tables and analyzed, and then poured into the results and discussion in this study, supported by related theories.

Results and discussion

1) Types of bacteria that play important role in the biofilter

Several type of bacteria are found in the biofilter reactor and most of them are known for their benefits and existence, for more details can be seen in Table 1.

The pollutants that appear in the table above are very harmful to the environment, and their pollution occurs in wastewater samples, biotrickling filters, and biofilms, which has encouraged the development of

processing methods to remove both organic matter and heavy metals. Biofiltration is an alternative method for removing organic matter and heavy metals [76]. This is because in the biofiltration reactor there is a medium where bacteria can grow and assist in the process of reducing existing pollutants [77]. In summary, pollutants present in liquid waste pass through the media [78] and are degraded by existing microorganisms [79, 80]. Microorganisms that grow on media degrade and develop biofilms. Good media have good surface area [81–82], water retention for biofilms to live on [83], and pollutant homogeneity [84]. These conditions provide space for microorganisms to reduce pollutants in liquid waste [47].

Table 1 Types of bacteria and their role in the biofilter reactor

No.	Bacterial type	Pollutans removal	Existence	References
1	<i>Comamonadacea</i>	Volatile fatty acids and aromatic compounds	Wastewater sample from biotrickling filter	[56]
2	<i>Proteobacteria, Bacteroidetes, and Actinobacteria</i>	Ammonium	More in biotrickling filter water samples than in biofilms	[57–58]
3	<i>Burkholderiales</i>	Ferric ions and hydrogen	-	[59]
4	<i>Comamonas testosteroni</i>	Nitrate	Biofilm	[58, 60]
5	<i>Comamonas nitrativorans</i>	Nitrate, nitrite and nitrous oxide	Wastewater sample from biotrickling filter	[61]
6	<i>Flavobacteriaceae, Alcaligenaceae, Cytophagaceae, Cryomorphaceae, Piscirickettsiaceae and Trueperaceae</i>	-	Wastewater sample from biotrickling filter	[62]
7	<i>Proteobacteria, Bacteroidetes dan Actinobacteria</i>	-	More in biotrickling filter water samples than in biofilms	[63–64]
8	<i>Pseudomonas taiwanensis, Acinetobacter guillouiae, Klebsiella pneumoniae</i>	Chromium, copper and zinc	Wastewater sample	[65]
9	<i>Azolla</i>	Strontium	Wastewater sample	[66]
10	<i>Rhodospirillum s sp.</i>	Cadmium, mercury, lead and nickel	Wastewater sample	[67]
11	<i>Eichhornia crassipes</i>	Lead, chromium, xinc, manganese and copper	Wastewater sample	[68]
12	<i>Gallionella ferruginea, Leptothrix sp.</i>	Arsenic and manganese	Wastewater sample	[69]
13	<i>Gracilaria sp.</i>	Aluminum, chromium and zinc	Wastewater sample	[70]
14	<i>Desulfovibrio sp.</i>	Zinc, copper, iron, arsenic and nickel	Wastewater sample	[71]
15	<i>Thiomonas sp.</i>	Arsenic and iron	Wastewater sample	[72]
16	<i>Thauera selenatis</i>	Zinc, copper, cadmium, nickel, mercury, lead, cobalt and chromium	Wastewater sample	[73]
17	<i>Sargassum filipendula</i>	Nickel and Copper	Wastewater sample	[74]
18	<i>Thiobacillum thiooxidans</i>	Zinc and copper	Wastewater sample	[75]

1.1) Organic pollutants removal

Comamonadacea can remove volatile fatty acid pollutants and aromatic compounds in a biotrickling filter [56] under both aerobic and anaerobic conditions. *Proteobacteria*, *Bacteroidetes*, and *Actinobacteria* are known to be able to remove ammonium, and their presence is higher in water samples than in biofilms [56–58]. In other studies, *Burkholderiales* were shown to be able to remove ferric ions, and it is also known that *Burkholderiales* have aerobic and anaerobic genera, and these bacteria are also able to oxidize hydrogen. Interestingly, these bacteria are also known as fermentative bacteria in the solution fermentation process [59]. In subsequent studies, it was found that the *Comamonas testosteroni* bacterium is present in biofilms and is known to be able to reduce nitrate; in general, these bacteria are abundant in biofilms, so the presence of these bacteria will accelerate nitrate reduction in wastewater [60]. In addition, it is also known that *Comamonas testosteroni* in bulk in an aerobic environment appears in both aerobic and anaerobic condition in biofilms [58].

Furthermore, *Comamonas nitrivorans* bacteria found in biotrickling filter water samples are known to reduce nitrite, nitrate and nitrous oxide [61]. Several bacteria, such as *Alcaligenaceae*, *Cryomorphaceae*, *Cytophagaceae*, *Piscirickettsiaceae*, *Flavobacteriaceae* and *Trueperaceae*, were identified to be present in the filter wastewater. Several bacteria exist in two places, namely biotrickling filters and biofilms, but the number is greater in biotrickling filters, including *Bacteroidetes*, *Proteobacteria*, and *Actinobacteria* [63–64]. *Alcaligenaceae*, *Cryomorphaceae*, *Cytophagaceae*, *Piscirickettsiaceae*, *Flavobacteriaceae*, and *Trueperaceae* bacteria in natural conditions have also been found in leachate waste in landfills [85–86].

1.2) Heavy metal pollutants removal

All heavy metals that enter the biofilter reactor are removed by biodegradation by bacteria previously attached to the biofilm [65, 87]. Several media have been found to reduce heavy metals because ion exchange occurs during biofiltration [88–89]. All heavy metals enter the biofilter reactor and are degraded by bacteria that grow on the biofilm to produce compounds that are insoluble and less toxic to the environment. This process is called adsorption, precipitation, bio-remediation, or absorption [90–92]. Heavy metals that pass through a biofilter reactor are converted into water and biomass, which can be safely discharged. Heavy metal methylation is another alternative that occurs in biofilter reactors, which are supported by bacteria present in biofilms [93–94].

Bacteria that can biodegrade heavy metals using biofiltration have been identified to be numerous, including *Pseudomonas taiwanensis*, *Acinetobacter guillouiae*, *Klebsiella pneumoniae* which are capable of reducing heavy metals of chromium, copper, and zinc types [65], *Azolla* which reduces strontium [66].

It is known also that *Rhodospirillum* sp. reduces mercury, cadmium, lead and nickel [67], *Thauera selenatis* reduces zinc, copper, cadmium, nickel, mercury, lead, cobalt and chromium [73], *Eichhornia crassipes* reduces chromium, manganese, lead, zinc and cadmium [68], *Leptothrix* sp. and *Gallionella ferruginea* reduce manganese and arsenic [69], *Sargassum filipendula* reduces copper and nickel [74], *Gracilaria* sp. reduces aluminum, zinc and chromium [70], *Thiobacillus thiooxidans* reduces zinc and copper [75], *Desulfovibrio* sp. reduces zinc, copper, iron, nickel and arsenic [71], *Thiomonas* sp. reduces arsenic and iron [72].

2) Several Factors affecting biofiltration effectiveness and efficiency

In a biofilter reactor, a complex procedure called biological filtration uses bacteria to break down the contaminants. Environmental factors affect biological aspects of wastewater remediation [95–97]. Temperature, dissolved oxygen, pH, and hydraulic retention time (HRT) have been measured in particular [98–101], nutrient supply, and biological organisms [102–103].

2.1) Temperature

By changing the temperature in the biofilter reactor, which impacts how well microorganisms develop on the formed biofilm, the performance of the biofilter can be controlled. Low temperatures are not ideal for the development of microorganisms, because some bacteria cannot endure them. Therefore, when low temperatures are present, bacteria that have already attached to the biofilm or are currently doing so will die [104–105]. Low temperatures are also known to contribute to long acclimatization [106] and low nitrification rates [107]. Additionally, it has been stated that 18°C is the ideal temperature for producing effective biofilter performance [107].

According to Brown et al. [108] temperature stimulation of biofilms has a significant effect on the metabolism and growth of bacteria. Thus, relatively warm temperatures may cause bacteria in the biomass and biofilms to multiply. Some studies have suggested that increasing temps may encourage the development of bacteria in biofilms [109–111]. Every living entity in the reactor experiences an increase in metabolism as the temperature increases. The composition of the bacteria residing in the reactor is known to vary with temperature [112–

113]. Ciliates are microorganisms that can thrive in reactors [114–115]. Instead of causing an increase in the quantity of microbial populations or the rate of photosynthesis, an increase in temperature may also result in a change in taxonomy [116–117]. Despite the notion that temperature is beneficial for the activity of photosynthetic enzymes [118–119], low nutrient amounts can prevent bacterial growth [120–121].

In the three-dimensional biofilm framework, the microbial communities were closely packed. The reaction of the biofilm to heat microbes adhering to the biofilm is modulated by the interactions between microbial groups [122–123]. Increasing temperatures can result in community changes when microbial species interact, which cannot be anticipated from the reactions of a single species [124]. Bacteria in biofilms fight one another for nutrients [125] while also taking advantage of one another [126].

2.2) Hydraulic retention time

The retention time must be considered if biofilter units are to be used over an extended period where retention duration has an impact on cost effectiveness as a direct result of how much substrate is handled [127–129]. Faster heterotrophic growth results from a prolonged retention time [130]. However, because it maintains a long retention time, extended retention also has some drawbacks such as cost consumption. [131].

According to [132–133], nitrite and nitrate effluents increase with a shorter hydraulic retention period. Based on the research conducted by [134], increased organic loading rate, hydraulic loading, and fluid shift as a result of hydraulic retention time aided bacterial development and biofilm formation. The bacterial community can degrade organic materials, and organic matter removal is more effective when the hydraulic retention time increases [135]. Thus, the reduction in organic matter during the pollutant degradation process can be influenced by the proper hydraulic retention time. A longer contact time between the wastewater and sedimentation rate of the treatment reactor can improve pollutant removal capacity [136–137].

2.3) Organic loading rate

Owing to the capacity of bacteria to disseminate organic processes beneath the flow of organic matter for the formation and refinement of biofilms that have been and will be created, the organic loading rate factor may have an effect on the rate at which biofilms grow [138]. The accumulation of biomass, biofilm development, and denitrification processes are all affected by the organic loading rate [139]. Excessive wastewater foam,

which prevents the development of biofilms, is a drawback of increasing the organic loading rate [138, 140–141] and decreasing the number of potential adsorption sites [142]. The gas formation rate remained constant as the organic loading rate increased, but the percentage of the effluent increased [143].

2.4) Dissolved oxygen

Dissolved oxygen is essential for cellular activity and pollutant biodegradation. The growth of the microbial community depended on the quantity of dissolved oxygen in the wastewater. This is due to the fact that many aerobic microorganisms rely on dissolved oxygen for life and growth [144]. Additionally, adequate air can promote biofilm development [145]. Low dissolved levels result in ineffective contaminant abatement [146].

As dissolved oxygen is used to oxidize ammonium, iron, and manganese, it has a major impact on the treatment of biological systems. Additionally, energy is typically required when oxygen is dissolved in water; as a result, the higher the dissolved oxygen in the water, the higher the energy intake that takes place. The increased energy of the biofilter reactor enables greater pollutant reduction [147].

2.5) Biological organisms

Biological organisms that serve as catalysts for biofiltration are the primary ingredients. Biofilms are frequently created by microorganisms, such as bacteria, protozoa, invertebrates, and fungi. Bacteria and fungi are thought to be beneficial for the growth of microbial populations in media. To complete the process of pollutant breakdown, heterotrophic microbes were immobilized. The cells initially adhered to the biofilter material on the reactor surface. Subsequently, microbial cells colonize the surface to create an active layer that can absorb pollutants [12]. Another method is to affix synthetic microorganisms to biofilter materials. Microencapsulation, membranes, cross-linking, carrier bonding, and trapping have been used to immobilize cells artificially [12]. These microorganisms grow on bioreactor substrates and are responsible for odor control and pollutant degradation [148–149].

2.6) Supply nutrient

Nutrition is crucial for assessing the effectiveness of biofiltration. Pollutants provide energy to microbes by serving as carbon sources. The primary macronutrients are potassium, sulfur, nitrogen, and phosphorus, while metals and vitamins constitute the majority of the micronutrients [102]. These minerals were added to the medium in either the liquid or solid form. In aqueous solutions, most mineral compounds are dissolved and

used as nutrient solutions in the biofilters. FeSO₄, CaCl₂, KNO₃, (NH₄)₂SO₄, KH₂PO₄, MnSO₄, MgSO₄, NH₄HCO₃, and NH₄Cl are frequently used mineral ions [150–151]. Numerous studies have shown that nutrient availability promotes microbial development [102, 152].

Conclusion

Thus, bacteria play an important role in biofiltration. Several types of bacteria were identified in the biofilter reactor in both biofilm and wastewater samples. In this process, bacteria are mobilized into the media and attached to the media so that the pollutants that pass through the media are broken down and degraded by microorganisms. These microorganisms behave like biocatalysts and also directly develop biofilms where they attach so that the biofilm will be wider, and the more area these microorganisms can occupy for growth. Some factors that affect the growth of biofilms also have a direct effect on the attached bacteria, such as the symbiosis of mutualism, where one organism and the other will need each other. Factors that support the symbiosis of mutualism between biofilms and bacteria include the type of media (area, size, and surface roughness), pollutant homogeneity, and water retention, which help biofilms to remain alive. In addition to helping in the process of organic pollutant degradation, these bacteria are known to help degrade heavy metals. The following types of bacteria can help biodegrade organic and heavy metal pollutants such as *Comamonadacea*, which can biodegrade volatile fatty acids and aromatic compounds. Proteobacteria, Bacteroidetes, and Actinobacteria can biodegrade ammonium. *Burkholderiales* can biodegrade ferric ions and hydrogen. *Comamonas testosteroni* is able to biodegrade nitrates. *Pseudomonas taiwanensis*, *Acinetobacter guillouiae*, and *Klebsiella pneumoniae*

can reduce copper, chromium and zinc levels. *Azolla* biomass reduces strontium. *Rhodospirillum* sp. can biodegrade cadmium, mercury, lead, and nickel. *Gallionella ferruginea* and *Leptothrix* sp. can biodegrade arsenic and manganese. *Gracilaria* sp. can biodegrade aluminum, chromium, and zinc. *Desulfovibrio* sp. can biodegrade copper, zinc, nickel, iron, and arsenic. *Thiomonas* sp. can biodegrade arsenic and iron. *Thauera selenatis* can biodegrade copper, zinc, cadmium, nickel, lead, cobalt, chromium, and mercury. *Thiobacillus thiooxidans* can degrade both zinc and copper. *Sargassum filipendula* biodegrades copper and nickel.

Meanwhile, in the findings of the factors that affected biofiltration, it was identified that there were four that played an important role such as temperature, dissolved oxygen, hydraulic retention time and organic loading rate, biological organisms, and supply nutrients. Temperature functions to control the performance of biofilms and bacteria in biofilms, where low temperatures will be very dangerous for the survival of bacteria that function to break down pollutants because it will reduce nutrient intake from wastewater for bacteria, so that they cannot develop and fall to the bottom of the bioreactor and are not replaced by other bacteria. Many bacteria that play an important role in biofilter reactors cannot survive at low temperatures. Retention time is also a key factor in biofilter performance because a longer contact time between bacteria and wastewater will further increase the number of bacteria that can adhere to the media, considering the roughness and breadth of the media. The rate of entry of the pollutant load is a key factor in biofilters because the rate of organic loading enhances the growth of existing bacterial biofilms. To summarize the conclusions, refer to the following Figure 1.

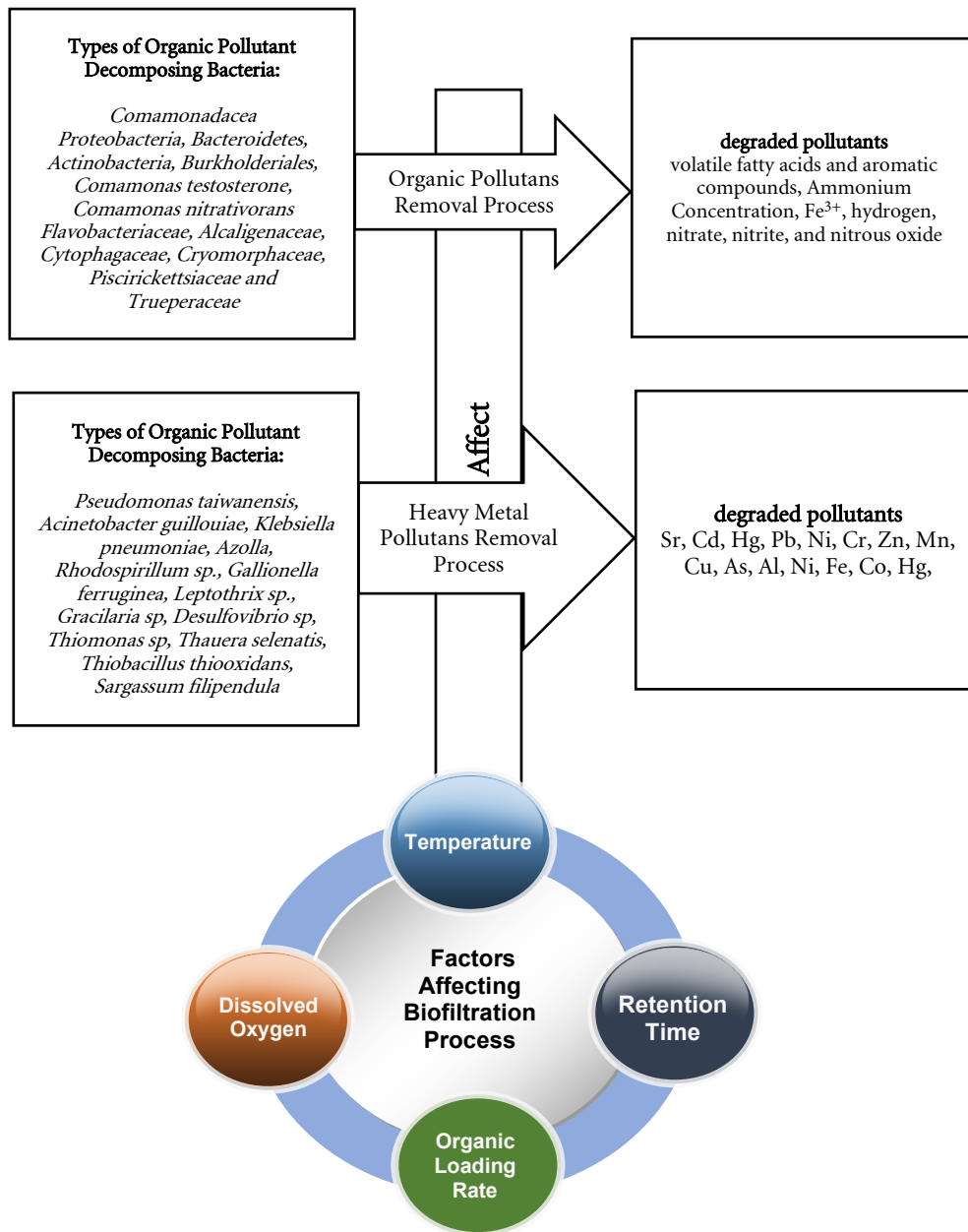


Figure 1 Summary conclusion.

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